## Charles’ Law Worksheet

Assume that the pressure and the amount of gas present remain constant in the following problems.

1. A gas at $89^{\circ} \mathrm{C}$ occupies a volume of 0.67 L . At what Celsius temperature will the volume increase to 1.12 L ?
2. The Celsius temperature of a 3.00 - L sample of gas is lowered from $80.0^{\circ} \mathrm{C}$ to $30.0^{\circ} \mathrm{C}$. What will be the resulting volume of this gas?
3. What is the volume of the air in a balloon that occupies 0.620 L at $25^{\circ} \mathrm{C}$ if the temperature is lowered to $0.00^{\circ} \mathrm{C}$ ?
